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Research Papers 2510 to 2558

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Corrections to be noted in volume 53 of the Journal of Research of the National Bureau of Standards

Page	Column	Line	Now reads in part	Should read
54	Fig. 3 legend		0.5 ml	0.05 ml
81	1	last	L. G. Overholser	[2] L. G. Overholser
82	Fig. 1 legend	4	or sample	for sample
86	2	eq (8)	$_0{}^oH$ (alpha)	H_{0^o} (alpha)
89	Figure 3		Douglas & Denver	Douglas & Dever
99	2	eq (3)	$+100[c_\vartheta - c_{et} - c_{pt}]$	$+100[c_\varphi - c_{et} - c_{pt}]$
99	2	19 from bottom	$c_\vartheta =$	$c_{pt} =$
102	Table 3, col. 3.		.001	.010
108	2	14 from bottom	C_2D_2H	C_2H_2D
141	Table 2	Footnote "c"	$[\Sigma d^2/(n-1)]^{\frac{1}{2}}$	$[\Sigma d^2/(n-1)]^{\frac{1}{4}}$
147	2	27	lower	higher
271	2	12	Add the sentence: This best average is obtained by dividing the sum of all three radioactive effects by the sum of all three masses.	
280	1	3	that pertaining for the	that of the
285	Table 1, col. 4.	{	β	$10^2\beta$
		{	1.74×10^2	1.74
287	Table 4, col. 3.	{	B	10^4B
		{	0.273×10^4	0.273
287	Table 4, col. 4.	{	C	10^7C
		{	0.27×10^7	0.27
366			Figure 6 legend delete: ———, Percentage of sample volatilized versus time; – – – –, logarithm to the base 10 of percentage residue versus time.	

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